

Subject Title: PHARMACEUTICAL CHEMISTRY-II

Subject Code: 0812

Important Instructions to examiners:

- 1) The answers should be examined by key words and not as word-to-word as given in the model answer scheme.
- 2) The model answer and the answer written by candidate may vary but the examiner may try to assess the understanding level of the candidate.
- 3) The language errors such as grammatical, spelling errors should not be given more Importance (Not applicable for subject English and Communication Skills.
- 4) While assessing figures, examiner may give credit for principal components indicated in the figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for anyequivalent figure drawn.
- 5) Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answers and model answer.
- 6) In case of some questions credit may be given by judgement on part of examiner of relevant answer based on candidate's understanding.
- 7) For programming language papers, credit may be given to any other program based on equivalent concept.



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1	d)	Define the following terms with example (any two)	
-	(,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	(i) Cardiotonic: The drugs or compounds which increase force of contraction without	
		increasing its oxygen consumption are known as cardiotonic e.g. Digitalis, stropanthus like	
		Digoxin Digitoxin Gitoxin	
		(ii) Vasodilator : These are the drugs which produces dilation of blood vessels by relaxing	
		smooth muscle cells, e.g. Hydralazine, Minoxidil, Nifedipine, Verapamil, Nitroglycerine,	
		Losartan, Prazosin, Doxazosin	1 M each
		(iii) Antidepressants Antidepressants are drugs which counteract or overcome mental	
		depression. These drugs are therapeutically useful in a variety of cases pertaining to mentally	
		ill patients. Mental depression is a phenomenon which may arise in normal individuals or in	
		mentally ill persons. E.g. Imipramine, Amitriptyline, Nortriptyline, Phenelzinesulphate,	
		Isocarboxid, Tranylcypromine, Mitrazapine, Trazodone	
1	e)	Give two brand names of following drugs (any two)	
		(i) Paracetamol: Tylenol, Calpol, panadol, crocin, metacin, valadol, paldesic, Dolo	1 M
		(ii) Metronidazole: Aristogyl, Flagyl, Metrogyl, Aldezol, Unimezol	each
		(iii) Salbutamol: Ashtalin, Respira, Salbetol, Ambrodil, Sobrex,, Salbuton, Asthasol	
1	f)	In what dosage form the following drugs are given (any two)	
		(i) Insulin :	
		1) Insulin Injection,	
		2) Insulin Injection Biphasic	
		3) Neutral Insulin Injection	
		4) Globin zinc Insulin Injection	
		5) Isophane Insulin Injection	1 M each
		6) Protamin zinc Insulin Injection	cach
		7) Insulin zinc Suspension	
		(ii) Mebendazole	
		1) Mebendazole Tablet	
		2) Mebensazole Syrup	
		(iii) Procaine : Procaine Injection	



MODEL ANSWER

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		3. Treatment of thyroid carcinoma.	
		4. Treatment of obesity.	
		5. It increases metabolism of carbohydrates, protein.	
		6. Rarely used in the treatment of male infertility and some gynaecological disorders.	
		7. It decreases serum cholesterol level.	
		(iii)Thrombin	
		1. Blood Coagulant.	
		2. Topically to control minor oozing due to superficial cuts.	
		3. Orally to prevent GIT bleeding.	
2		Attempt any <u>FOUR</u> of the following:	12M (4X3M)
2	a)	Classify Antimalarial drugs. Give structure of chloroquine	2 14
		Classification:	2 M Classifi
		a) Alkaloids – e.g. Quinine	cation,
		b) 4-amino quinolines – e.g. Chloroquine, Amodiaquine	1M-
		c) 8-amino quinolines – e.g. Primaquine	Structu
		d) 9- aminoAcridine : e.g. Mepacrine	re-
		e) Biguanides – e.g. Proguanil	
		f) Pyrimidines – e.g. Pyrimethamine, Trimethoprim	
		g) Miscellaneous – e.g Protonsil, Dapsone, Artesunate, Artemether etc.	
		Structure of Chloroquine :	
		CH ₂ CH ₂ -CH ₂	
		NH-CH-CH ₂ -CH ₂ -CH ₂ -CH ₂	
		CH ₂ -CH ₃	
2	b)	Nome one drug used for .	1 M
2	U)	(i) Condidiogia: Nustatin Amphatariain P. Elucopazola Katacopazola Itracopazola	each
		(1) Canuluiasis Nystaun, Amphoteneni-D, Fluconazole, Ketoconazole, Ilfaconazole,	
		Ciou illiazoie.	



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		(ii) Amoebiasis: Emetine, Clioquinol, Diiodohydroxyquinoline, Metronidazole, Tinidazole,	
		Ornidazole, Carbarsone, Diloxanide furoate, Paramomycin, Erythromycin.	
		(iii) Leprosy : Dapsone, Rifampicin, Clofazimine, Thiambutosine, Solapsone, Thiacetazone.	
2	c)	Classify Adrenergic drugs. Draw structure of any one Catecholamine.	2 M
		The adrenergic drugs can be classified based on their chemical structure.	Classifi cation.
		1) Catecholamines e.g : Adrenaline, Nor-adrenaline, Isoprenaline	1M-
		2) Non-Catecholamines e.g. Phenylephrine, Salbutamol, Terbutaline, Ephedrine,	Structu re-
		Pseudoephedrine.	
		3) Imidazoline derivatives eg. Naphazoline, Tetrahydrozolium	
		OR	
		1. Vasoconstrictors († B. P.): Noradrenaline (Norepinephrine), Dopamine, Ephedrine etc.	
		2. Cardiac stimulants: Dopamine, Adrenaline, Isoprenaline	
		3. CNS stimulants: Amphetamine	
		4. Smooth muscle relaxants: Adrenaline, Isoprenaline, salbutamol etc.	
		5. Drugs used in allergic reactions: Ephedrine	
		6. Local vasoconstrictor/ nasal decongestants: Phenylephrine, pseudoephedrine	
		7. Anorectics: Amphetamine, Phentermine.	
		Catecholamine: (Any one Structure will carry ONE mark)	
		Adrenaline	
		OR	



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		Properties:	
		1. It occurs as an odorless, white crystalline powder.	
		2. It is very slightly soluble in water, freely soluble in alcohol.	
		3. It is dextrorotatory.	
		Uses of testosterone :	
		1. It has both androgenic and anabolic activity. Its primary use is as androgen replacement	
		therapy in men at maturity age in case of testosterone deficiency.	
		2. It is useful in certain anemias, osteoporosis and to stimulate growth in undergrown boys.	
		3. It is used to increase athletic performance and maintain muscle tone.	
		4. Used in palliative treatment of disseminated breast cancer in postmenopausal women.	
		5. Used in treatment of gynaecomastia.	
2	e)	Define diuretics? Give any one method of classification for diuretics with example.	1M-
		Diuretics: Drugs which promote excretion of water & electrolytes from body through kidneys in	Define, 2 M
		the form of urine are called diuretics.	Classifi
		Classification:-	cation,
		1) Water & Osmotic agents	
		a) Electrolytes:-Sodium & Potassium salts	
		b) Non electrolytes:- Mannitol, Urea	
		2) Organic mercurials:- Mersalyl acid	
		3) Acidifying agents:-Ammonium chloride, Arginine hydrochloride	
		4) Alpha-beta unsaturated ketones:- Ethacrynic acid	
		5) Purinase & related compound: Caffeine	
		6) Sulphonamides:-	
		a) Carbonic anhydrase inhibitors-e.g. Acetazolamide	
		b) Benzothidiazines: - Chlorthiazide, Hydrochlorthiazide	
		c) Sulphamoyl benzoic acid derivatives e.g. Frusemide	
		7) Endocrine antagonists: (aldostrone antagonists) e.g. Spironolactone	
		8) Miscellaneous agents: - Trimaterene	



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		OR	
		Diuretics can also be classified as	
		1) Weak diuretics –	
		a) Osmotic diuretics:- Sodium & Potassium salts	
		b) Xanthine deri.:- Aminophylline	
		c) Carbonic anhydrase inhibitors-e.g. Acetazolamide	
		2) Moderatly efficacious diuretics: -	
		a) Osmotic diuretics: - Mannitol, Sucrose, Glycerol	
		b) Benzothiadiazines deri Chlorthalidone, Chloroxozone	
		3) Very efficacious diuretics (High celling diuretics) e.g. Frusemide & Ethacrynic acid	
		4) Potassium sparing diuretics:	
		a) Aldostrone antagonists: - Spironolactone	
		b) Renal epithelial sodium channel inhibitors: - Trimaterene, Amiloride	
2	f)	What are antihistaminics? Give classification of antihistaminics with examples.	
		An antihistaminic is an agent that inhibits the release or action of histamine and can be used to	1M- Meanin
		describe any histamine antagonist, but it is usually reserved for the classical antihistamines that	g
		act upon the H1 histamine receptor and H2 receptor blockers are used in the treatment of	2 M
		stomach ulcer, gastric ulcer, heart burn etc.	Classifi
		Classification of Antihistaminics:	cation,
		1. H1 blockers or H1 antagonist:	
		a. Aminoalkylethers/Ethanolamines e.g. Diphenhydramine, Doxylamine	
		b. Ethylenediamine e.g. Mepyramine, Tripelennamine, Pyrilamine	
		c. Alkylamines/Propylamines e.g. Pheniramine, Chlorpheniramine, Triprolidine	
		d. Phenothiazine derivatives e.g. Promethazine, Trimeprazine	
		e. Piperazine derivatives. e.g Meclizine, Cyclizine, Chlorcyclizine	
		f. Dibenzocycloheptenes: Cyproheptadine, Azatadine	
		g. Second generation antihistaminics: e.g. Cetrizine, Levocetrizine, Fexofenadine, Terfenadine	
		2. H2 Blockers or H2 receptor antagonist e.g. Ranitidine, Cimetidine, Famotidine	
		3. An inhibitor of histamine release e. e.g. Sodium Cromoglycate	



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3		Attempt any <u>FOUR</u> of the following:	(12 M) (4x3M)
3	a)	Write structure, Give chemical name, properties, and uses of Caffeine? Structure of Caffeine: H_3C N N N N N N N N	1 M
		Chemical name :- 1.3.7 trimethyl xanthine	1 M
		Properties:	
		1. It occurs as white crystalline powder having bitter taste.	0 5 M
		2. It sublimes on heating.	0.5 11
		3. Sparingly soluble in water but very soluble in boiling water.	
		4. It is a very weak base.	
		Uses:	
		1. Stimulation of central nervous system.	0.5 M
		2. Used as diuretic.	
		3. Vasodilation of peripheral vessels.	
		4. Decreases drowsiness.	
		5. Relieve mental fatigue and headache of certain kind like neuralgia, rheumatism, migraine etc.	
3	b)	Name any two halogenated hydroxyl Quinolines. Draw structure and Give chemical name of	
		DEC?	
		Halogenated hydroxyl Quinolines:	
		Iodoquinol (Di-iodohydroxyquinoline),	1 M
		clioquinol(chloroiodoquinol),	each
		Cloxyquin (chlorohydroxyquinoline)	



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	Structure: $\begin{array}{c} & & & \\ & &$	
3 c)	Name the two barbiturates used as "General anaesthetics". Draw structure and Give chemical name of Phenobarbitone. Barbiturates used as General anaesthetics: Methohexitone, Thiopentone sodium Structure: Chemical name: 5-ethyl, 5-phenyl barbituric acid.	1 M each



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3	d)	Give structure, chemical name and storage condition of Aspirin? Structure:	1 M each
		<u>Chemical name</u> : Acetyl salicylic acid <u>Storage conditions:</u> It should be stored in air tight containers, in a cool, dry place.	
3	e)	What are narcotic analgesics? Give classification of narcotics analgesic with examples.	
		Narcotic analgesics	
		Narcotic analgesics are derivatives of opium, semi synthetic or synthetic agents having potent	1 M
		analgesic & narcotic activity and effective for the treatment of severe pain.	
		Classification of Narcotic analgesics:	
		Narcotic analgesic are classified as:-	2 M
		1. Morphine and related compounds (Natural alkaloids of opium) e.g. Morphine, Codeine.	
		2. Semi-synthetic derivatives of morphine- Heroin, Brown Sugar	
		3. Synthetic Agents- Methadone, Pethidine, Dextropropoxyphen hydrochloride.	
		OR	
		Classification of Narcotic analgesics:	
		1. Naturally occurring:	
		a) Morphine and it's analogues: e.g.: Morphine, Codeine	
		2. Synthetic:	
		a) Morphinan analogues: e.g.: Levorphenol	
		b) Benzomorphan analogues: e.g.: Pentazocin	
		c) 4-Phenylpiperidine analogues: e.g.: Pethidine	



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3 f) W Vi Cl		
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	What are vitamins? Give classification of vitamins with examples.	
01	Vitamins:	
01	Vitamins may be defined as potent organic substances which are essential for normal growth	1M
01	and maintenance of life of animals, which they are not able to synthesize in adequate quantity	
01	and their deficiency may cause various diseases.	
01	Classification:	
0	1. Fat soluble vitamins:	2 M
0	E.g.: Vitamin A (Retinol), Vitamin D (Calciferol), Vitamin E (Tocopherol), Vitamin K	
0	(Phytomenadione)	
O	2. Water-soluble vitamins:	
O	E.g.: Vitamin B1 (Thiamine), Vitamin B2 (Riboflavin / Lactoflavin), Vitamin B6	
0	(Pyridoxine), Vitamin B12 (Cyanocobalamin), Folic acid, Nicotinic acid, Vitamin C	
0	(Ascorbic acid)	
0	3. Fat- water insoluble vitamin:	
0	E.g.: Vitamin H (Biotin)	
	OR	
	1. Fat soluble vitamins:	
	a) Are obtained from β-ionone ring: e.g.: Vitamin-A	
	b) Are obtained from steroids/sterol: e.g.: Vitamin-D	
	c) Contain chromane ring: e.g.: Vitamin-E	
	d) Contain naphthaquinone ring: e.g.: Vitamin-K	
	2. Water-soluble vitamins:	
	E.g.: Vitamin B1 (Thiamine), Vitamin B2 (Riboflavin / Lactoflavin), Vitamin B6	
	(Pyridoxine), Vitamin B12 (Cyanocobalamin), Folic acid, Nicotinic acid, Vitamin C	
	(Ascorbic acid)	



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4		Attempt any <u>FOUR</u> of the following:	(12 M) (4x3M)
4	a)	Classify of Antibiotics with examples	
		I. β-Lactam antibiotics:	3 M
		e.g. Benzyl Penicillin, Phenoxymethyl penicillin, Cephaloridine, cephalothin	
		II. Non-β-Lactam antibiotics:	
		1. Tetracyclines: e.g chlortetracycline, oxytetracycline.	
		2. Aminoglycoside antibiotics : e.g: Streptomycin, neomycin, gentamicin	
		3. Macrolide antibiotics : e.g : Erythromicin	
		4. Ansamycins : e.g: Rifamycin	
		5. Polyene macrolide antibiotics: e.g: Nystatin, Hamycin	
		6. Anthracycline antibiotics : e.g :actinomycin, daunorubicin	
		7. Peptide antibiotics: e.g: Bacitracin.	
		8. Steroidal antibiotics : e.g : Fusidic acid	
		9. Nucleoside anitibiotics: e.g : Puromycin	
		10. Non- classifiable antibiotics : e.g : Chloramphenicol	
4	b)	Explain the terms "Lipid Lowering Agent". Give properties and uses of Clofibrate.	
		Lipid lowering agents:	
		Hyperlipidemia is the most prevalent indicator for susceptibility to atherosclerotic heart disease	1 M
		& it also describes elevated plasma levels of lipids that are usually in the form of lipoproteins.	each
		Drugs which are used to reduce the elevated levels of the lipids in the blood are called Lipid	
		lowering agents.	
		Properties:	
		1. It is a clear, almost colorless liquid.	
		2. It has a characteristic odor.	
		3. It is having acrid taste first and then becomes sweet.	
		4. It is very slightly miscible in water and miscible in alcohol.	
		5. It is heat stable.	



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		Uses:	
		1. It is used in the treatment of type III hyperlipoproteinaemia.	
		2. It is used in the treatment of severe hypertriglyceridemia.	
		3. It is also used in long term treatment and prophylaxis of coronary heart disease.	
4	c)	Define anti-neoplastic drugs. Write uses of cyclophosphamide and methotrexate.	
		Definition:	
		Anti-neoplastic agents, also known as Cytotoxic agents are used in the treatment of malignant	1 M
		diseases, when surgery or radiotherapy is not possible or has proved ineffective.	each
		Uses of Cyclophosphamide:	
		1. Used in treatment of solid tumours such as carcinoma of the breast, cervix, lung and ovary.	
		2. Used in combination of other agents in the treatment of lymphomas, myeloma.	
		3. Used as immunosuppressant in tissue and organ transplantation.	
		4. Used in the management of autoimmune disorders such as nephritic syndrome and	
		rheumatoid arthritis.	
		Uses of Methotrexate:	
		1. Used in the management of acute lymphoblastic leukemia.	
		2. Used as immunosuppressant.	
		3. Given by mouth or by injection as methotrexate sodium.	
4	d)	Define 'Parasympathomimetics'. Give properties and uses of Pilocarpine.	
		Parasympathomimetics:	
		The drug which exert or mimic the pharmacological action / effects of acetylcholine or drugs	
		which bring about stimulation of parasympathetic nervous system are called	1 M
		parasympathomimetics.	each
		Properties of Pilocarpine:	
		1. Pilocarpine is colourless crystals or a white crystalline powder.	
		2. It is odourless.	
		3. It is sensitive to light.	
		Uses of Pilocarpine: It is used :	
		i) As miotic.	
		ii) To reduce intraocular pressure in glaucoma.	



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		iii) For diagnosis of Adie's (tonic) pupil.	
		iv) To counteract anticholinergic side effects (eg. Dryness of mouth, constipation etc).	
4	e)	Define Diabetes Mellitus? Discuss storage, precautions and labelling of Insulin Preparation.	
		Diabetes Mellitus:	
		Diabetes mellitus is a metabolic disorder characterized by hyperglycaemia usually associated	1M
		with polyphagia, polydypsia; polyuria, glycosuria, weight loss, dehydration etc. caused due to	
		deficiency or diminished effectiveness of insulin (insulin resistance).	
		Storage condition:	
		All insulin preparations must be stored at low temperatures between 2-8°C in a dark place.	1 M
		Precautions:	0 5 M
		Insulin is affected by heat, light and moisture so protect from it.	0.5 11
		In case of multi-dose container should be shaken gently before withdrawal of dose.	
		Labelling: The label should bear-	0 5 M
		a) number of unit per ml	0.0 111
		b) the animal source of insulin	
		c) expiry date	
		d) do not freeze	
4	f)	Give structure, chemical name, properties and uses of Indomethacin.	
		Structure:	
		H ₃ CO H ₃ CO CH ₂ COOH CH ₃ CH ₃	1M
		Chemical Name: 1-(p-chlorobenzoyl)5-methoxy 2-methyl indol-3yl-acetic acid	1 M
		Properties of Indomethacin:	0.535
		1. It occurs as a pale yellow to brownish yellow crystalline powder.	0.5 M



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I		2. It is a declarge and almost to stale ==	
		2. It is odoriess and almost tasteless.	
		3. It is very slightly soluble in water and sparingly soluble in alcohol.	
		4. It is stable in neutral or slightly acidic media.	0514
		5. It is decomposed by strong alkali and sunlight.	0.5 M
		Uses of Indomethacin:	
		It is used as Analgesic, Anti-inflammatory and Antipyretics for the treatment of –	
		1. Rheumatoid arthritis	
		2. Acute gout	
		3. Spondylitis	
		4. Dysmenorrhea	
		5. Acute musculo-skeletal disorder	
		6. Pain in malignant disease	
5		Attempt any <u>FOUR</u> of the following	12M
			(4X3M)
5	a)	What are Cardiovascular drugs? Classify them with examples.	
		Definition	1 M
		• Cardiovascular agents include various types of drugs having an action on the heart or on other	
		parts of the vascular system and they have the ability to alter cardiovascular function.	
		<u>OR</u>	
		• Cardiovascular Agents represents a group of drugs which have direct action on the heart or	
		other parts of the vascular system so that they modify the total output to the heart or the	
		distribution of blood to certain parts of the circulatory system.	
		Classification of cardiovascular agents:-	
		Different kinds of drugs fall under this category like:	2M
		1) Cardiotonics (Positive cardiac inotropic agents):- they increase the force of contraction of the	
		myocardium e.g. Cardiac glycosides obtained from Digitalis, Stropanthus, squill such as	
		Digoxin, Digitoxin, Lanatoside C etc.	
		2) Antiarrhythmic drugs:- used to regulate arrhythmic (irregular) contraction of cardiac muscles	
		of the heart. eg. Quinidine, Procainamide, Phenytoin, lignocaine hydrochloride,	
		propranalol etc.	



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		It is used in the treatment of following diseases:	1M
		Uses:	
		Structural modifications are possible	
		Inactivated by enzyme <i>penicillinase</i> and gastric juice	
		• Degraded rapidly in strong acidic and basic media	
		• Very soluble in water	
		Hygroscopic, Dextrorotatory	0.5M
		• White, finely crystalline powder with faint characteristic odour	
		Properties:	
		6-(2-phenyl ethanoylamino) 2,2-dimethyl penam-3-carboxylic acid.	0.5M
		Chemical name: 6-(2-phenyl acetamido) penicillanic acid. OR	
		Structure : $Ph - H_2C - C - HN - CH_3 - C$	1M
5	b)	Write structure, give chemical name, properties and uses of Penicillin G.	
		c) Antiplatelet drugs eg. Aspirin	
		b) Diuretics eg. Furosemide, Hydrochlorthiazide	
		a) Anticoagulants eg. Warfarin, Dicoumarol	
		6) Other drugs which indirectly affect cardiovascular system:	
		5) Antihyperlipidemic agents: (lipid lowering agents) e.g Clofibrate, Nicotinic acid	
		Propranalol, Atenolol, Prazosin, Tolazoline, Hydralazine, Minoxidil, Verapamil, Captopril etc.	
		pressure. e.g. α-methyldopa, clonidine, Pentolinium, Mecamylamine, Reserpine, Guanethidine,	
		4) Anti-hypertensive:-which regulate the blood pressure by decreasing the elevated blood	
		e.g. Amyl nitrate, Isosorbid nitrate, Verapamil, Propranolol	
		meet its metabolic demands for oxygen.	
		3) Antianginal agents:-which are used in the treatment of angina pectoris, enabling the heart to	



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1) Respiratory tract infection 2) Urinary tract infection 3) Gonorrhea 4) Syphilis 5) Meningitis 6) Enteric infection 7) Septicemia. 8) Abscesses Prophylactically used before dental and surgical procedures to prevent from developing endocarditis and re-occurrence of rheumatic fever. 5 What are tranquilizers? Write structure, give chemical name and popular trade name of c) Chlorpromazine. **Tranquilizers: -**1MTranquillizers are CNS depressants which bring about a calming effect and induce a mild sedative effect. These are the agents or drugs which reduce anxiety, induce mental repose, and suppress agitation without significantly diminishing mental alacrity, they may cause some drowsiness but tolerance soon develops to this effect. 1M**Structure:** H_2 H₂ CH₂ 0.5M **Chemical name:** 2-chloro-10-(3-dimethylaminopropyl)phenothiazine Trade names: (any one) Largactil, Chlorozine, Copamide, Chlorectil plus, Chlorzen plus, Clozine **0.5M**



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5	d)	Name one biguanide derivative used as hypoglycemic agent. Write its structure and uses.	1M
		Following biguanide derivatives are used as hypoglycemic agent.	each
		Phenformin, Metformin	
		Structure of Phenformin:	
		$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	
		Uses of phenformin:	
		• To treat non-insulin dependent diabetes mellitus	
		• To reduce blood sugar level in cortisone induced hyperglycemia	
		• To reduce blood cholesterol in maturity onset diabetes.	
5	e)	Write structure of Propantheline bromide, give its chemical name, properties and uses.	
		Structure:	1M
		H_3C CH_3 O O N^+ CH_3 CH_3 CH_3 H_3C CH_3 CH_3 CH_3 Br^-	
		Chemical name:	0.5M
		N,N-di-isopropyl-N-methyl-N-[2-(xanthene-9-yl carbonyloxy)ethyl]ammonium bromide	
		Properties:	0.5M
		• It occurs as white or yellowish white powder, odorless and has very bitter taste	
		• Slightly hygroscopic and soluble in water	1M
		Uses:	
		• To treat gastric and duodenal ulcers.	
		To treat intestinal hypermotility.	



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 To reduce gastric secretion. To produce reactive hypoglycemia (by stimulating insulin release). To reduce biliary and uterine spasm. To control excessive sweating and salivation. To prevent nocturnal enuresis in children. To prevent nocturnal enuresis in children. Following drugs are used as antithyroid drugs: Propylthiouracil, carbimazol, methimazole, methylthiourocil. Structure of thyroxine: HO I HO I Attempt any FOUR of the following a) Write the name of the microorganism which is responsible for huma structure, give chemical name, properties and uses of DDS. 	1M each
• To produce reactive hypoglycemia (by stimulating insulin release).• To reduce biliary and uterine spasm.• To control excessive sweating and salivation.• To prevent nocturnal enuresis in children.5 f) 5 f) Name two antithyroid drugs. Draw structure of thyroxine.Following drugs are used as antithyroid drugs: Propylthiouracil, carbimazol, methimazole, methylthiourocil.Structure of thyroxine: $HO_{++++} + f(+++) + f(++++) + f(+++++) + f(+++++) + f(+++++) + f(+++++) + f(++++++) + f(++++++) + f(+++++++) + f(+++++++) + f(++++++++) + f(++++++++++$	1M each
 For reduce biliary and uterine spasm. To control excessive sweating and salivation. To prevent nocturnal enuresis in children. 1 Name two antithyroid drugs. Draw structure of thyroxine. Following drugs are used as antithyroid drugs: Propylthiouracil, carbimazol, methimazole, methylthiourocil. Structure of thyroxine: HO, f, f, I, f, f, L, f, f,	1M each
 For control excessive sweating and salivation. To prevent nocturnal enuresis in children. To prevent nocturnal enuresis in children. Name two antithyroid drugs. Draw structure of thyroxine. Following drugs are used as antithyroid drugs: Propylthiouracil, carbimazol, methimazole, methylthiourocil. Structure of thyroxine: HO I HO I HO I Attempt any FOUR of the following a) Write the name of the microorganism which is responsible for huma structure, give chemical name, properties and uses of DDS. 	1M each
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 Propylthiouracil, carbimazol, methimazole, methylthiourocil. Structure of thyroxine: HO I HO I <lii< li=""> <lii< li=""> I</lii<></lii<>	each
 Structure of thyroxine: HO, I, I,	
Image: Hole of the following Hole of the following Attempt any FOUR of the following Attempt any FOUR of the microorganism which is responsible for human structure, give chemical name, properties and uses of DDS.	
6Attempt any FOUR of the following6a)Write the name of the microorganism which is responsible for huma structure, give chemical name, properties and uses of DDS.	
6 a) Write the name of the microorganism which is responsible for huma structure, give chemical name, properties and uses of DDS.	16M
6a)Write the name of the microorganism which is responsible for huma structure, give chemical name, properties and uses of DDS.	(4X4M)
structure, give chemical name, properties and uses of DDS.	n Leprosy. Write
Leprosy is caused by slow growing bacteria, Mycobacterium Leprae	0.5M
Structure of DDS (Dapsone)	1 M
Chemical name:	
Bis (4-aminophenyl) sulphone or 4,4'-diamino, diphenyl sulphone	1M
Properties:It is white or slightly white crystalline powder.	



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	• It is odorless.	
	• It is bitter in taste, practically insoluble in water, soluble in alcohol, freely soluble in acetone	
	and dilute mineral acids.	
	Uses-	
	• Dapsone (diamino-diphenyl sulfone) is a pharmacological medication most commonly used	1M
	in combination with rifampicin and clofazimine as multidrug therapy (MDT) for the	
	treatment of Mycobacterium leprae infections (leprosy).	
	• Dapsone is used in combination with pyrimethamine in the treatment of malaria.	
	• It is also used in the treatment of dermatitis herpetiformis and relapsing polychondritis	
	• In combination with trimethoprim or pyrimethamine it is used to treat pneumonia.	
6 b)	Write structure and give chemical name of PAS and Pyrazinamide.	1M
	Structure of PAS:	each
	$f_{H_2N} = f_{H_2N} + f_{H_2N} $	



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Subject

Write structure, give chemical name, properties and uses of Furosemide. Structure: $CI \longrightarrow CH_2 \longrightarrow CH_2 \longrightarrow CH_2 \longrightarrow COOH$ Chemical name: 4-chloro-N-furfuryl-5-sulphamoyl anthranilic acid OR	1M each
$C + CH_2 + CH_2 + CH_2 + CH_2 + COOH$ Chemical name: 4-chloro-N-furfuryl-5-sulphamoyl anthranilic acid OR	1M each
Chemical name: 4-chloro-N-furfuryl-5-sulphamoyl anthranilic acid OR	
4-chloro-N-furfuryl-5-sulphamoyl anthranilic acid OR	
4-chloro-2-furfuralamino-5-sulphamoyl benzoic acid	
Properties:	
• It is white crystalline powder, odorless, tasteless,	
• Very slightly soluble in water but soluble in solution of alkali hydroxides	
Uses:	
• It is used as diuretic	
• To treat oedema associated with congestive heart failure, liver cirrhosis and renal diseases	
• For management of hypertension	
Draw structure of Pyrimethamine. Give its properties, storage conditions and pharmaceutical	1M
uses.	each
Structure:	
$\begin{array}{c} CI \\ \\ \\ H_3C \\ \\ \\ H_2 \end{array} \\ \\ \\ H_2 \end{array} \\ \\ \\ NH_2 \\ \\ \\ NH_2 \\ \\ \\ NH_2 \\ \\ \\ \\ NH_2 \\ \\ \\ \\ NH_2 \\ \\ \\ \\ \\ NH_2 \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	
]	• For management of hypertension Draw structure of Pyrimethamine. Give its properties, storage conditions and pharmaceutical ises. Structure: $ \begin{bmatrix} CI \\ $



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Properties: It is white crystalline powder. Odourless, tasteless. Practically insoluble in water and soluble in warm dilute mineral acids. **Storage conditions:** It is affected by light and hence it is stored in tightly closed light resistant container **Pharmaceutical uses:** It is used for prophylaxis and treatment of malaria. In combination with sulphadiazine it is used to treat toxoplasmosis. What are general anaesthetics? Classify them with examples. Draw structure of cyclopropane. 6 e) **1M Definition**: General anaesthetics are the central nervous system depressant drugs which bring about loss of all modalities of sensations along with a reversible loss of consciousness. **Classification:** $2\mathbf{M}$ 1) Inhalation anaesthetics: which include the liquids of volatile nature and gaseous substances used by inhalation to produce anaesthesia. These may be sub-classified as follows: Volatile liquids: i. e.g. Chloroform. Trichloroethylene, a) Halogenated hydrocarbons: Halothane. Ethylchloride b) Ethers : e.g. Diethyl ether, Vinyl ether ii. Gases: e.g. Cyclopropane, Nitrous oxide 2) Intraveneous anaesthetics:-Barbiturates: Ultra short acting barbiturates such as Methohexitone, Thiopentone sodium i. ii. Non-barbituates: a) Eugenol derivatives. e.g. Propanidid b) Phencyclidine derivatives. e.g Ketamine c) Steroids. e.g. Althesin



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Miscellaneous such as Etomidate, Propofol. **Structure of cyclopropane:** H_{2} **1M** $H_{2}C$ ·CH₂ 6 What are 'Local Anaesthetics'? Write structure, give chemical name of local Anaesthetic drug f) having following chemical feature. i) Ester ii) Amide **Definition: 1M** Local anaesthetics are drugs which produce insensitivity in a limited area around the site of application or injection of the drug by preventing generation and conduction of impulses along nerve fibres and nerve ending and the effects are reversible. Structure of drug having 1.5M i) Ester : (Procaine) each H₂N C₂H₅ Chemical name: 4-amino-(2-diethyl amino ethyl) benzoate or 2-(Diethyl amino) ethyl-4-amino benzoate.



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